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Synthesis and biological evaluation of tricyclic anilinopyrimidines as $IKK\beta$ inhibitors

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ABSTRACT

A series of tricyclic anilinopyrimidines were synthesized and evaluated as IKK β inhibitors. Several analogues, including tricyclic phenyl (**10**, **18a**, **18c**, **18d**, and **18j**) and thienyl (**26** and **28**) derivatives were shown to have good in vitro enzyme potency and excellent cellular activity. Pharmaceutical profiling of a select group of tricyclic compounds compared to the non-tricyclic analogues suggested that in some cases, the improved cellular activity may be due to increased $clog\ P$ and permeability.

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The nuclear factor-κB (NF-κB) family of transcription factors plays an important role in many biological processes, including inflammatory immune responses, oncogenesis, tumor metastasis, and chemo resistance.¹⁻⁶ NF-κB is normally retained in the cytoplasm as an inactive form associated with the IkB inhibitory proteins. 7 However, upon cellular stimulation, $I\kappa B\alpha$ is phosphorylated by the IkB kinase (IKK) and subsequently undergoes ubiquitin-dependent degradation. This liberates NF-κB from the IκB/NFκB complex allowing its entry into the nucleus where it regulates the transcription of an array of genes responsible for cell survival and growth.⁸⁻¹⁰ The IkB kinase (IKK) complex is composed of at least two catalytic subunits IKKα, IKKβ, and a noncatalytic regulatory protein, IKKγ, also known as IKK-1, IKK-2, and NEMO, respectively. 11-13 Although both catalytic subunits are capable of phosphorylating IκBα, IKKβ has been identified as essential for the activation of the complex in response to inflammatory stimuli. 14,15 The potential for utilizing IKKβ inhibitors for the treatment of cancer and immunological disorders has been widely recognized by the pharmaceutical industry as demonstrated by the abundance of recent publications^{16–23} and patents.^{24,25}

Our investigations to identify an IKK β inhibitor began with a high throughput screening of compounds through an IKK β Lance assay.²⁶ The studies led to the discovery of a series of compounds containing an anilinopyrimidine core, such as that contained in **1**

(Fig. 1).²⁷ These analogues, including the lead compound **1**, had an excellent in vitro IKK β inhibitory profile. However, further evaluations of the scaffold revealed less impressive activity in cellular assays^{28,29} potentially due to poor physical properties. Research efforts were then directed toward improving both the cellular activity and aqueous solubility of these analogues in an effort to produce potent IKK β inhibitors with good pharmacokinetic profiles.

Initial SAR studies revealed that substitution of the sulfonamide of ${\bf 1}$ with water solubilizing groups led to compounds with similar in vitro IKK β inhibitory activity, but improved aqueous solubility and pharmacokinetic properties. ³⁰ Additionally, we found that substitution of the anilinopyrimidine in the 4-position with alternative heteroaryl or aryl moieties produced compounds with similar biological profiles. In an effort to further develop our understanding of the SAR of this series, it was decided to examine

Me
$$\stackrel{S}{\longrightarrow} \stackrel{N}{\stackrel{N}{\mapsto}} \stackrel{N}{\stackrel{N}{\mapsto}} SO_2NH_2$$

1

IKKβ IC₅₀ = 0.019 ± 0.016 μM

IKKβ IC₅₀ = 0.019 ± 0.016 μM IKKβ Reporter IC₅₀ = 1.71 ± 0.75 μM Cell Growth (LoVo) IC₅₀ = 1.66 ± 0.85 μM Aq. Solubility = 3.0 μg/mL

Figure 1. Lead compound from exploratory investigations.

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Scheme 1. Reagents and conditions: (a) DMF-DMA, 110-125 °C, 24 h; (b) 3.0 equiv 4, NMP, 105-120 °C, 24 h to 3 d.

Table 1 $\mbox{IKK}\beta$ inhibition of non-tricyclic and corresponding tricyclic analogues

Compound	Core	R	IKKβ IC ₅₀ ^a (μM)	IKKβ Reporter IC ₅₀ ^b (μM)	Cell growth (LoVo) IC ₅₀ ^c (μM)
6	I	Н	20.16 ^d	nd ^e	3.28
7	II	Н	1.83	1.43	1.05
8	III	Н	>10,000	4.06 ^f	4.18 ^g
9	I	OMe	0.04 ^h	0.64 ⁱ	2.55
10	II	OMe	0.55	0.42	0.75

- ^a [ATP] = 2 μ M; N = 1, unless otherwise noted; see Ref. 26.
- b N = 1, unless otherwise noted: see Ref. 28.
- c N = 1, unless otherwise noted; see Ref. 29.

sired anilinopyrimidine derivatives.

- ^d N = 2, (SD ± 1.68 μ M).
- e nd = not determined.
- f N = 2, (SD ± 1.84 μ M).
- g N = 2, (SD ± 0.16 μ M).
- h N = 7, (SD ± 0.016 μ M).
- i N = 2, (SD ± 0.16 μ M).
- substitution of the pyrimidine in more detail. One of these investigations was focused on developing a set of anilinopyrimidines with bridged heteroaryl or aryl substitution (e.g., tricyclic compound 5) to probe the effects of restricted rotation. These tricyclic compounds were produced using a route similar to that employed for the synthesis of non-bridged analogues of this series (Scheme 1). Reaction of ketone 2 with DMF-DMA led to enamine 3, which cy-

clized under thermal conditions with guanidine 4 to form the de-

The IKK β activity of the tricyclic phenyl compounds compared to their corresponding non-tricyclic analogues is outlined in Table 1. It was found that 6–6–6 tricyclic systems (Table 1, core II) served as suitable cores, however, 6–7–6 tricyclic systems (Table 1, core III) were not tolerated. While these tricyclic compounds had moderate enzyme activity, a significant improvement of cellular activity in both the reporter and cell growth assays³¹ was observed.

Additional SAR investigations revealed an alternative approach for improving cellular activity. Substitution of the phenyl ring *para* to the pyrimidine with acylamino groups had led to compounds with excellent enzyme potency and cellular activity. Compounds such as **11**, **12**, and **13** had approximately 10-fold increase in cell growth potency³¹ over other anilinopyrimidine analogues (Fig. 2). This led us to investigate IKK β inhibitory activity of tricyclic derivatives of these acylamino compounds.

We focused on developing a small library of acylamino-phenyl substituted tricyclic analogues (18). Utilizing a route similar to that described previously, reaction of the amino-substituted tetralone 14 with DMF-DMA formed dienamino compound 15, which was condensed with substituted guanidines 4 to form the tricyclic anilinopyrimidine unit 16. Addition of acyl chlorides led to the formation of the 4-acylamino-phenyl tricyclic analogues 18 (Scheme 2).

Figure 2. Acylamino-phenyl substituted anilinopyrimidines.

Scheme 2. Reagents and conditions: (a) DMF-DMA, 120 °C, 24 h; (b) 1.3 equiv 4, NMP, 120 °C, 24 h, then 2 N aq NaOH, 100 °C; (c) 1.2 equiv 17, 2.5 equiv diethyl aniline, NMP, rt, 24 h.

Table 2 IKKβ inhibition of acylamino-phenyl substituted tricyclic anilinopyrimidines

Compound	R	IKKβ IC_{50}^{a} (μM)	IKKβ Reporter IC_{50}^{b} (μ M)	Cell growth (LoVo) IC ₅₀ ^c (μM) 0.79	
11	_	0.01	2.46 ^d		
18a	Neopentyl	0.82	1.36	0.37	
12		0.05	4.47	0.18	
18b	Benzyl	2.38	1.79	0.19	
13	-	0.02 ^e	6.91 ^f	0.25 ^g	
18c	Thiophen-2-ylmethyl	0.51	0.83	0.29	
18d	Thiophen-3-ylmethyl	0.31	0.97	0.49	
18e	4-Methoxybenzyl	1.83	1.60	0.73	
18f	isobutyl	1.94	6.40	0.38	
18g	2-Methylprop-1-enyl	1.59	8.25	0.56	
18h	Cyclopentyl	2.20	1.57	0.83	
18i	Isopropyl	1.11	1.46	0.66	
18j	3-Methoxy-3-oxo-propionyl	0.28	nd ^h	0.62	
18k	2-Furyl	2.23	nd ^h	1.06	

[[]ATP] = 2 μ M; N = 1, unless otherwise noted; see Ref. 26.

NH
$$SO_2NH_2$$

NH A

SO₂NH A

SO₂

Scheme 3. Reagents and conditions: (a) DMF-DMA, 120–125 °C, 24 h to 6 d; (b) 3 equiv **4**, NMP, 140 °C, 24 h; (c) 3.0 equiv. guanidine carbonate, NMP, 115 °C, 3 d; (d) 1.2 equiv **23**, 15 mol % Pd₂dba₃, 30 mol % BINAP, 3.5 equiv NaOt-Bu, dioxane, NMP, microwave, 115 °C, 1–2 h.

N = 1, unless otherwise noted; see Ref. 28.

 $^{^{}c}$ N = 1, unless otherwise noted; see Ref. 29.

^d N = 2, (SD ± 0.49 μ M).

 $^{^{}e}$ N = 2, (SD ± 0.006 μ M).

N = 2, (SD \pm 0.000 μ M). N = 3, (SD \pm 1.69 μ M). N = 4, (SD \pm 0.24 μ M).

h nd = not determined.

Table 3 IKK β inhibition of non-tricyclic and corresponding tricyclic thiophene analogues

Compound	Core	R	IKKβ IC ₅₀ ^a (μM)	IKKβ Reporter IC ₅₀ ^b (μM)	Cell growth (LoVo) IC ₅₀ ^c (μM)
25	IV	Н	0.18	1.70	2.62
26	V	Н	0.79	0.80	0.33
27	IV	2-Dimethylaminoethyl	0.23	0.77	2.81
28	V	2-Dimethylaminoethyl	0.77	0.51	0.68
29	IV	3-Dimethylaminopropyl	0.17	1.06	3.38
30	V	3-Dimethylaminopropyl	0.78	0.62	1.10
31	IV	Ethanol-2yl	0.13	3.39	4.11
32	V	Ethanol-2yl	1.22	1.96	0.80

^a [ATP] = 2 μ M; N = 1; see Ref. 26.

A diverse set of targets containing aryl- and heteroaryl-substituted acylamino groups were synthesized and evaluated for IKK β activity (Table 2). Several compounds exhibited potent cellular activity in the reporter assay and modest enzyme activity. However, some analogues demonstrated potency in both assays. In particular, **18a**, **18c**, and **18d** and **18j** had excellent cellular potency and good IKK β enzyme activity.

In addition to investigating phenyl tricyclic derivatives, we also explored tricyclic thiophenes. In this case, we examined both the effect of bridging the aniline pyrimidine substitution and the incorporation of water solubilizing groups substituted at the sulfonamide. As determined from previous SAR studies, substitution of the sulfonamide with amino- or hydroxyl-alkyl side chains re-

tains activity while improving aqueous solubility and pharmacokinetic properties.³⁰ We hoped to produce tricyclic analogues that would have improved cellular activity and pharmaceutical profile.

The scheme below outlines the synthesis of both non-tricyclic and tricyclic thiophene analogues. The cyclocondensation route described in Schemes 1 and 2 was used to produce thiophene analogues without substitution of the sulfonamide (21). To incorporate water solubilizing groups, it was necessary to synthesize 2-amino-substituted pyrimidine 22, which underwent microwave-assisted Suzuki couplings with a variety of aryl bromides that contained substituted sulfonamides to form the desired analogues 24 (Scheme 3).

Table 4 IKK β inhibition and pharmaceutical profiling of non-tricyclic and corresponding tricyclic analogues

Compound	Core	IKKβ $IC_{50}^{a}(\mu M)$	IKKβ Reporter IC ₅₀ ^b (μM)	Cell growth (Lovo) IC ₅₀ ^c (μM)	clog P	Permeability $^{\rm d}$ (Pe \times 10^{-6} cm/ $_{\rm S})$	Solubility pH: 7.4° (μg/ mL)
6	I	20.16	nd ^f	3.28	2.79	2.26	1.0
7	II	1.83	1.43	1.05	3.24	2.92	7.0
9	I	0.04	0.64	2.55	2.83	<0.01	2.3
10	II	0.55	0.42	0.75	3.28	1.80	<1.0
11	I	0.01	2.46	0.79	3.88	0.60	<1.0
18a	II	0.82	1.36	0.37	4.34	2.30	9.0
12	I	0.05	4.47	0.18	3.57	<0.01	1.0
18b	II	2.38	1.79	0.19	4.02	3.37	9.0
13	I	0.02	6.91	0.25	3.21	0.30	<1.0
18c	II	0.51	0.83	0.29	3.67	0.51	6.0
25	IV	0.18	1.70	2.62	2.48	0.50	2.0
26	V	0.79	0.80	0.33	2.93	0.38	<1.0
27	IV	0.23	0.77	2.81	3.49	2.40	>100
28	V	0.77	0.51	0.68	3.94	1.54	36.0

^a [ATP] = 2 μ M; N = 1, unless noted in previous tables; see Ref. 26.

^b *N* = 1; see Ref. 28.

c N = 1; see Ref. 29.

^b N = 1, unless noted in previous tables; see Ref. 28.

 $^{^{\}rm c}$ N = 1, unless noted in previous tables; see Ref. 29.

d Effective permeability via passive diffusion determined using Parallel Artificial Membrane Permeability Assay (PAMPA).

e Aqueous solubility.

f nd = not determined.

The IKK β activity of the tricyclic thiophene compounds compared to their corresponding non-tricyclic analogues is shown in Table 3. Similar to the tricyclic phenyl compounds, the thiophene derivatives also exhibited improved cellular activity in the reporter assay, but a decrease in IKK β enzyme inhibition. Additionally, incorporation of water solubilizing groups had small effects on the in vitro IKK β activity of these tricyclic compounds.

The pharmaceutical properties of a select group of tricyclic compounds and their corresponding non-tricyclic counterparts were evaluated. In general, the tricyclic phenyl compounds (Table 4, core II) had increased cellular permeability, as predicted by the clog *P* values. However, the tricyclic thiophene derivatives (Table 4, core V) did not have the same permeability profile. Despite having higher clog *P* values than their corresponding non-tricyclic analogues, the thiophene tricyclics did not have improved permeability. Further investigation using Caco-2 analysis confirmed a decrease in permeability for these analogues.³² This suggested the importance of closely monitoring off target activity or modes of cellular transport when pursuing the compounds. In addition, the incorporation of a water solubilizing groups greatly increased the compounds solubility and potentially pharmacokinetic properties, which would be of importance for the design of future analogues.

In summary, we have reported several tricyclic anilinopyrimidine derivatives that inhibit IKK β . These tricyclic compounds have increased cellular potency over the corresponding non-tricyclic analogues. In general, the improved cellular activity may be due to the compounds' increased $clog\ P$ value and permeability profile. However, this relationship was not observed in a select group of thiophene analogues, indicating the possibility that biological activities in addition to IKK β inhibition are contributing to the observed cellular activity.

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- 31. Inhibition of cell growth may be to off target activity.
- 32. Caco-2 Cell-Layer Permeability Assay (Papp × 10⁻⁶ cm/s): **25** (a-b) 53.08, (b-a) 38.89; **26** (a-b) 33.95, (b-a) 20.68; **27** (a-b) <1.00, (b-a) 13.16; **28** (a-b) <1.00, (b-a) 3.08.